NEW BRUNSWICK CHEMISTRY 121-122 · UNIT C · 5 QUESTIONS

Acids and Bases — Practice Test

This practice test was made by AI and there may be mistakes. If a question or answer looks wrong, use the Flag a mistake button to let Brett know so he can fix it.

1

A solution has hydrogen ion concentration 0.0001 moles per litre. Its pH is



ANSWER

Understand It with Pulse

Flag a mistake

2 **A solution has pH 6. Its hydrogen ion concentration in moles per litre is**

ANSWER _____

Understand It with Pulse | Flag a mistake

3 **A solution has pH 10. Its pOH is**

ANSWER _____

Understand It with Pulse | Flag a mistake

4 **A solution has pOH 2. Its pH is**

ANSWER _____

Understand It with Pulse | Flag a mistake

5 **A 0.01 molar strong acid fully dissociates. Its hydrogen ion concentration in moles per litre is**

ANSWER _____

Understand It with Pulse | Flag a mistake

ANSWER KEY

1. 4
2. 0
3. 4
4. 12
5. 0.01

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